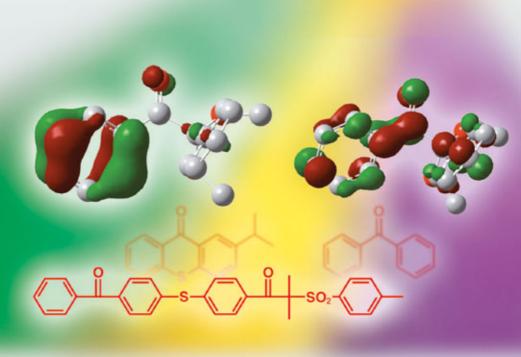
Photoinitiators for Polymer Synthesis

Scope, Reactivity, and Efficiency



Jean Pierre Fouassier and Jacques Lalevée

Photoinitiators for Polymer Synthesis

Related Titles

Ramamurthy, V.

Supramolecular Photochemistry Controlling Photochemical Processes

2011

ISBN: 978-0-470-23053-4

Allen, N. S. (ed.)

Handbook of Photochemistry and Photophysics of Polymeric Materials

2010

ISBN: 978-0-470-13796-3

Albini, A., Fagnoni, M. (eds.)

Handbook of Synthetic Photochemistry

2010

ISBN: 978-3-527-32391-3

Wardle, B.

Principles and Applications of Photochemistry

2010

ISBN: 978-0-470-01493-6

Stochel, G., Stasicka, Z., Brindell, M., Macyk, W., Szacilowski, K.

Bioinorganic Photochemistry

2009

ISBN: 978-1-4051-6172-5

Jean Pierre Fouassier and Jacques Lalevée

Photoinitiators for Polymer Synthesis

Scope, Reactivity and Efficiency



WILEY-VCH Verlag GmbH & Co. KGaA

The Authors

Prof. Jean Pierre Fouassier

formerly University of Haute Alsace Ecole Nationale Supérieure de Chimie 3 rue Alfred Werner 68093 Mulhouse Cedex France

Prof. Jacques Lalevée

University of Haute Alsace Institut Science des Matériaux IS2M-LRC 7228, CNRS 15 rue Jean Starcky 68057 Mulhouse Cedex France All books published by Wiley-VCH are carefully produced. Nevertheless, authors, editors, and publisher do not warrant the information contained in these books, including this book, to be free of errors. Readers are advised to keep in mind that statements, data, illustrations, procedural details or other items may inadvertently be inaccurate.

Library of Congress Card No.: applied for

British Library Cataloguing-in-Publication Data

A catalogue record for this book is available from the British Library.

Bibliographic information published by the Deutsche Nationalbibliothek

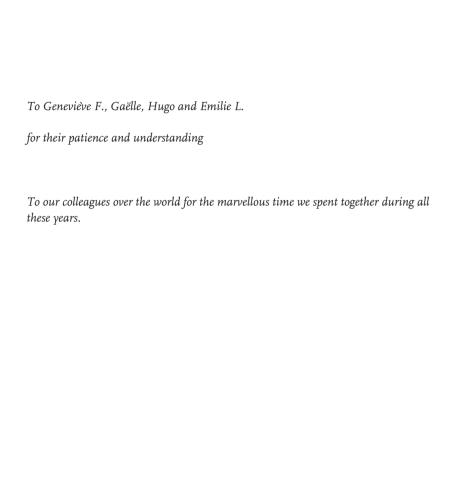
The Deutsche Nationalbibliothek lists this publication in the Deutsche Nationalbibliografie; detailed bibliographic data are available on the Internet at http://dnb.d-nb.de>.

© 2012 Wiley-VCH Verlag & Co. KGaA, Boschstr. 12, 69469 Weinheim, Germany

All rights reserved (including those of translation into other languages). No part of this book may be reproduced in any form – by photoprinting, microfilm, or any other means – nor transmitted or translated into a machine language without written permission from the publishers. Registered names, trademarks, etc. used in this book, even when not specifically marked as such, are not to be considered unprotected by law.

Cover Design Adam-Design, Weinheim
Typesetting Laserwords Private Limited,
Chennai, India
Printing and Binding Markono Print
Media Pte Ltd, Singapore

Print ISBN: 978-3-527-33210-6 ePDF ISBN: 978-3-527-64827-6 ePub ISBN: 978-3-527-64826-9 mobi ISBN: 978-3-527-64825-2 oBook ISBN: 978-3-527-64824-5



Contents

	Abbreviations XIX
	Introduction XXV
	Part I Basic Principles and Applications of Photopolymerization Reactions 1
1	Photopolymerization and Photo-Cross-Linking 3 References 6
2	Light Sources 11
2.1	Electromagnetic Radiation 11
2.2	Characteristics of a Light Source 12
2.3	Conventional and Unconventional Light Sources 13
2.3.1	Xenon Lamp 13
2.3.2	Mercury Arc Lamp 14
2.3.3	Doped Lamps 14
2.3.4	Microwave Lamps 14
2.3.5	Excimer Lamps 14
2.3.6	Light-Emitting Diodes (LEDs) 16
2.3.7	Pulsed Light Sources 17
2.3.8	Laser Sources 17
2.3.9	Sun 18
2.3.10	Household Lamps 18
2.3.11	UV Plasma Source 19
	References 20
3	Experimental Devices and Examples of Applications 21
3.1	UV Curing Area: Coatings, Inks, Varnishes, Paints, and
	Adhesives 21
3.1.1	Equipment 21
3.1.2	End Uses 21

VIII	Contents	
	3.1.3	Coating Properties 24
	3.2	Conventional Printing Plates 25
	3.3	Manufacture of Objects and Composites 25
	3.4	Stereolithography 25
	3.5	Applications in Microelectronics 26
	3.6	Laser Direct Imaging 26
	3.7	Computer-to-Plate Technology 27
	3.8	Holography 27
	3.9	Optics 28
	3.10	Medical Applications 28
	3.11	Fabrication of Nano-Objects through a Two-Photon Absorption Polymerization 29
	3.12	Photopolymerization Using Near-Field Optical Techniques 29
	3.13	Search for New Properties and New End Uses 30
	3.14	Photopolymerization and Nanotechnology 32
	3.15	Search for a Green Chemistry 33
		References 34
	4	Photopolymerization Reactions 41
	4.1	Encountered Reactions, Media, and Experimental Conditions 41
	4.2	Typical Characteristics of Selected Photopolymerization
		Reactions 45
	4.2.1	Film Radical Photopolymerization of Acrylates 45
	4.2.2	Film Cationic Photopolymerization 48
	4.2.3	Thiol-ene Photopolymerization 49
	4.2.4	Photopolymerization of Water-Borne Light Curable Systems 52
	4.2.5	Photopolymerization of Powder Formulations 52
	4.2.6	Charge-Transfer Photopolymerization 53
	4.2.7	Dual Cure Photopolymerization 54
	4.2.8	Hybrid Cure Photopolymerization 55
	4.2.9	Anionic Photopolymerization 55
	4.2.10	Metathesis Photopolymerization 56
	4.2.11	Controlled Photopolymerization Reactions 56
	4.2.11.1	Radical Photopolymerization Reactions 56
	4.2.11.2	Cationic Photopolymerization Reactions 58
	4.2.11.3	Anionic Photopolymerization Reactions 59
	4.2.12	Hybrid Sol–Gel Photopolymerization 59
	4.2.13	Photo-Cross-Linking Reactions in the Presence of Photobases or
		Photoacids 59
	4.3	Two-Photon Absorption-Induced Polymerization 60
	4.4	Remote Curing: Photopolymerization without Light 60
	4.5	Photoactivated Hydrosilylation Reactions 61
		References 61

5	Photosensitive Systems 73
5.1	General Properties 73
5.2	Absorption of Light by a Molecule 74
5.2.1	Absorption 74
5.2.2	Molecular Orbitals and Energy Levels 74
5.2.3	Absorption of Light and Optical Transitions 74
5.2.4	Reciprocity Law 76
5.2.5	Multiphotonic Absorption 77
5.3	Jablonski's Diagram 78
5.4	Kinetics of the Excited State Processes 78
5.5	Photoinitiator and Photosensitizer 80
5.6	Absorption of a Photosensitive System 81
5.7	Initiation Step of a Photoinduced Polymerization 82
5.7.1	Production of Initiating Species 82
5.7.2	Competitive Reactions in the Excited States 83
5.7.3	Reactivity in Bulk versus Solution: Role of the Diffusion 84
5.7.4	Cage Effects 85
5.8	Reactivity of a Photosensitive System 86
	References 87
6	Approach of the Photochemical and Chemical Reactivity 89
6.1	Analysis of the Excited-State Processes 89
6.1.1	Nanosecond Laser Flash Photolysis 89
6.1.2	Picosecond Pump-Probe Spectroscopy 90
6.1.3	Photothermal Techniques 90
6.1.4	Time-Resolved FTIR Spectroscopy 91
6.1.5	Direct Detection of Radicals 92
6.1.6	CIDNP, CIDEP, and ESR Spectroscopy 92
6.2	Quantum Mechanical Calculations 93
6.3	Cleavage Process 94
6.4	Hydrogen Transfer Processes 95
6.5	Energy Transfer 96
6.6	Reactivity of Radicals 98
	References 99
7	Efficiency of a Photopolymerization Reaction 103
7.1	Kinetic Laws 103
7.1.1	Radical Photopolymerization 103
7.1.2	Cationic Photopolymerization 105
7.1.3	Dependence of the Photopolymerization Rate 106
7.1.4	Laser-Induced Photopolymerization 107
7.1.5	Kinetics of the Photopolymerization in Bulk 108
7.2	Monitoring the Photopolymerization Reaction 109
7.2.1	FTIR Analysis 109
7.2.2	Photocalorimetry 110

х	Contents	
	7.2.3	Optical Pyrometry 110
	7.2.4	Other Methods 110
	7.3	Efficiency versus Reactivity 111
	7.4	Absorption of Light by a Pigment 112
	7.5	Oxygen Inhibition 114
	7.6	Absorption of Light Stabilizers 115
	7.7	Role of the Environment 117
		References 118
		Part II Radical Photoinitiating Systems 123
	8	One-Component Photoinitiating Systems 127
	8.1	Benzoyl-Chromophore-Based Photoinitiators 127
	8.1.1	Benzoin Derivatives 129
	8.1.2	Benzoin Ether Derivatives 132
	8.1.3	Halogenated Ketones 136
	8.1.4	Dialkoxyacetophenones and Diphenylacetophenones 136
	8.1.5	Morpholino and Amino Ketones 137
	8.1.6	Hydroxy Alkyl Acetophenones 137
	8.1.7	Ketone Sulfonic Esters 140
	8.1.8	Thiobenzoate Derivatives 140
	8.1.9	Sulfonyl Ketones 144
	8.1.10	Oxysulfonyl Ketones 147
	8.1.11	Oxime Esters 148
	8.2	Substituted Benzoyl-Chromophore-Based Photoinitiators 148
	8.2.1	Benzoin Ether Series 149
	8.2.2	Morpholino Ketone and Amino Ketone Series 149
	8.2.3	Hydroxy Alkyl Acetophenone Series 151
	8.2.3.1	Oil- or Water-Soluble Compounds 151
	8.2.3.2	Difunctional Compounds 152
	8.2.4	Modified Sulfonyl Ketones 154
	8.2.5	Limit of the Substituent Effect 154
	8.2.6	Macrophotoinitiators 156
	8.2.7	Supported Cleavable Photoinitiators 157
	8.3	Hydroxy Alkyl Heterocyclic Ketones 157
	8.4	Hydroxy Alkyl Conjugated Ketones 158
	8.5	Benzophenone- and Thioxanthone-Moiety-Based Cleavable
		Systems 158
	8.5.1	Benzophenone Phenyl Sulfides 158
	8.5.2	Ketosulfoxides 159
	8.5.3	Benzophenone Thiobenzoates 159
	8.5.4	Benzophenone-Sulfonyl Ketones 159
	8.5.5	Halogenated Derivatives 160

8.5.6	Cleavable Benzophenone, Xanthone, and Thioxanthone Derivatives 160
8.6	Benzoyl Phosphine Oxide Derivatives 161
8.6.1	Compounds 161
8.6.2	Excited State Processes 162
8.6.3	Absorption Properties and Photolysis 164
8.6.4	Bis-acyl phosphine oxide/Phenolic Compound Interaction 165
8.7	Phosphine Oxide Derivatives 165
8.8	Trichloromethyl Triazines 165
8.9	Biradical-Generating Ketones 166
8.10	Peroxides 166
8.11	Diketones 167
8.12	Azides and Aromatic Bis-Azides 168
8.13	Azo Derivatives 168
8.14	Disulfide Derivatives 168
8.15	Disilane Derivatives 169
8.16	Diselenide and Diphenylditelluride Derivatives 170
8.17	Digermane and Distannane Derivatives 170
8.18	Carbon–Germanium Cleavable-Bond-Based Derivatives 170
8.19	Carbon–Silicon and Germanium–Silicon Cleavable–Bond-Based
0.17	Derivatives 172
8.20	Silicon Chemistry and Conventional Cleavable Photoinitiators 172
8.21	Sulfur–Carbon Cleavable-Bond-Based Derivatives 173
8.22	Sulfur–Silicon Cleavable-Bond-Based Derivatives 173
8.23	Peresters 173
8.24	Barton's Ester Derivatives 174
8.25	Hydroxamic and Thiohydroxamic Acids and Esters 174
8.26	Organoborates 176
8.27	Organometallic Compounds 176
8.27.1	Titanocenes 177
8.27.2	Chromium Complexes 177
8.27.3	Aluminate Complexes 178
8.28	Metal Salts and Metallic Salt Complexes 178
8.29	Metal-Releasing Compound 178
8.30	Cleavable Photoinitiators in Living Polymerization 179
8.30.1	Cleavable C–S- or S–S-Bond-Based Photoiniferters 179
8.30.2	TEMPO-Based Alkoxyamines 182
8.31	Oxyamines 183
8.31.1	Alkoxyamines 183
8.31.2	Silyloxyamines 184
8.32	Cleavable Photoinitiators for Two-Photon Absorption 184
8.33	Nanoparticle-Formation-Mediated Cleavable Photoinitiators 185
8.34	Miscellaneous Systems 185
8.35	Tentatively Explored UV-Light-Cleavable Bonds 185
	References 187

9	Two-Component Photoinitiating Systems 199
9.1	Ketone-/Hydrogen-Donor-Based Systems 199
9.1.1	Basic Mechanisms 199
9.1.2	Hydrogen Donors 201
9.1.2.1	Amines 201
9.1.2.2	Thio Derivatives 203
9.1.2.3	Benzoxazines 204
9.1.2.4	Aldehydes 204
9.1.2.5	Acetals 204
9.1.2.6	Hydroperoxides 205
9.1.2.7	Silanes 205
9.1.2.8	Silylamines 207
9.1.2.9	Metal-(IV) and Amine-Containing Structures 208
9.1.2.10	Silyloxyamines 210
9.1.2.11	Germanes and Stannanes 210
9.1.2.12	Phosphorus-Containing Compounds 211
9.1.2.13	Borane Complexes 211
9.1.2.14	Alkoxyamines 212
9.1.2.15	Monomers 213
9.1.2.16	Photoinitiator Itself 214
9.1.2.17	Alcohols and THF 214
9.1.2.18	Polymer Substrate 214
9.1.2.19	Silicon-Hydride-Terminated Surface 214
9.1.3	Benzophenone Derivatives 215
9.1.3.1	Benzophenone 215
9.1.3.2	Aminobenzophenones 218
9.1.3.3	Other Benzophenones 218
9.1.3.4	Photopolymerization Activity 219
9.1.4	Thioxanthone Derivatives 219
9.1.4.1	Thioxanthone 219
9.1.4.2	Substituted Thioxanthones 220
9.1.5	Diketones 223
9.1.5.1	Aromatic Diketones 223
9.1.5.2	Camphorquinone 224
9.1.6	Ketocoumarins 225
9.1.7	Coumarins 225
9.1.8	Alkylphenylglyoxylates 225
9.1.9	Other Type II Ketone Skeletons 228
9.1.9.1	Anthraquinones 228
9.1.9.2	Fluorenones 229
9.1.10	Aldehydes 229
9.1.11	Aliphatic Ketones 229
9.1.12	Cleavable Ketones as Type II Photoinitiators 229
9.1.13	Tailor-Made Type II Ketones 229

9.1.13.1	Low- and High-Molecular-Weight Compounds and	
0 1 12 2	Macrophotoinitiators 229	
9.1.13.2	Water-Soluble Compounds 233	
9.1.13.3	Two-Photon Absorption Photoinitiators 235	
9.1.13.4	Photomasked Photoinitiator 235	
9.1.13.5	Oxygen Self-Consuming Thioxanthone Derivatives 236	
9.1.13.6	Low-Molecular-Weight One-Component Systems 236	
9.2	Dye-Based Systems 238	
9.2.1	Dye/Amine Systems 238	
9.2.2	Dye/Coinitiator Systems 239	
9.2.3	Improvement of Dye/Amine Systems 240	
9.2.4	Dye/Amine Water-Soluble Systems 241	
9.2.5	Kinetic Data 241	
9.3	Other Type II Photoinitiating Systems 241	
9.3.1	Maleimide/Amine and Photoinitiator/Maleimide 241	
9.3.2	Donor/Acceptor Systems 243	
9.3.3	Bisarylimidazole Derivative/Additive 244	
9.3.4	Pyrylium and Thiopyrylium Salts/Additive 245	
9.3.5	Ketone/Ketone-Based Systems 246	
9.3.6	Organometallic Compound/Ketone-Based Systems 247	
9.3.7	Organometallic Complex/Additive 248	
9.3.7.1	Organometallic Derivatives 248	
9.3.7.2	Ferrocenium Salts 248	
9.3.8	Metal Carbonyl/Silane 250	
9.3.9	Photosensitizer-Linked Photoinitiator or Coinitiator-Based	
	Systems 250	
9.3.10	Photoinitiator/Peroxide- or Hydroperoxide-Based Systems	251
9.3.11		252
9.3.12	Photoinitiator/Phosphorus-Containing Compounds 253	
9.3.13	Photoinitiator/Onium Salts 254	
9.3.14	Photosensitizer/Titanocenes 254	
9.3.15	•	255
9.3.16	,	256
9.3.17	Miscellaneous Two-Component Systems 257	
9.3.17.1	Light Absorbing Amine/Monomer 257	
9.3.17.2	Nitro Compound/Amine 257	
9.3.17.3	Hydrocarbon/Amine 257	
9.3.17.4	Photosensitizer/Triazine Derivative 257	
9.3.17.5	Other Systems 258	
7.3.17.3	References 258	
10	Multicomponent Photoinitiating Systems 269	
10.1	Generally Encountered Mechanism 269	
10.1.1	Ketone/Amine/Onium Salt or Bromo Compound 269	
10.1.2	Dye/Amine/Onium Salt or Bromo Compound 270	

XIV	Contents	
	10.1.3	Ketone/Amine/Imide Derivatives 270
	10.1.4	Ketone/Ketone/Amine 271
	10.1.5	Dye/Amine/Triazine Derivative 271
	10.2	Other Mechanisms 271
	10.2.1	Triazine-Derivative-Containing Three-Component Systems 272
	10.2.2	Dye/Ketone/Amine 272
	10.2.3	Dye/Amine/Metal Salt 272
	10.2.4	Dye/Borate/Additive 273
	10.2.5	Photosensitizer/Cl-HABI/Additive 273
	10.2.6	Metal Carbonyl Compound/Silane/Hydroperoxide 275
	10.2.7	Photosensitizer/Amine/HABI Derivatives/Onium Salt 276
	10.2.8	Dye/Ferrocenium Salt/Hydroperoxide 277
	10.2.9	Coumarin/Amine/Ferrocenium Salt 277
	10.2.10	Dye/Ferrocenium Salt/Amine/Hydroperoxide 278
	10.2.11	Organometallic Compound/Silane/Iodonium Salt 279
	10.3	Type II Photoinitiator/Silane: Search for New Properties 279
	10.4	Miscellaneous Multicomponent Systems 281
		References 281
	11	Other Photoinitiating Systems 283
	11.1	Photoinitiator-Free Systems or Self-Initiating Monomers 283
	11.2	Semiconductor Nanoparticles 284
	11.3	Self-Assembled Photoinitiator Monolayers 284
		References 285
		Part III Nonradical Photoinitiating Systems 287
	12	Cationic Photoinitiating Systems 289
	12.1	Diazonium Salts 289
	12.2	Onium Salts 289
	12.2.1	Iodonium and Sulfonium Salts 289
	12.2.1.1	Compounds 289
	12.2.1.2	Photopolymerization Reaction 290
	12.2.1.3	Role of the Anion 292
	12.2.1.4	Absorption Properties 293
	12.2.1.5	Decomposition Processes of Iodonium Salts 294
	12.2.1.6	Decomposition Processes of Sulfonium Salts 299
	12.2.1.7	Acylsulfonium Salts 301
	12.2.1.8	Substituted Iodonium and Sulfonium Salt Derivatives 301
	12.2.2	Other Onium Salts 303
	12.2.3	Search for New Properties 304
	12.2.3.1	Absorption 304
	12.2.3.2	Solubility 305 Stability 305
	12.2.3.3	,
	12.2.3.4	Benzene Release 307

12.2.3.5	Odor 307
12.2.3.6	Toxicity 307
12.2.3.7	Amphifunctionality 308
12.3	Organometallic Derivatives 308
12.3.1	Transition Organometallic Complexes 308
12.3.2	Inorganic Transition Metal Complexes 310
12.3.3	Non-Transition-Metal Complexes 310
12.4	Onium Salt/Photosensitizer Systems 311
12.4.1	Photosensitization through Energy Transfer 312
12.4.2	Photosensitization through Electron Transfer 312
12.4.2.1	Photosensitizers 312
12.4.2.2	Photoinitiation Step 317
12.5	Free-Radical-Promoted Cationic Photopolymerization 318
12.5.1	Radical/Onium Salt Interaction 318
12.5.2	Radical Source/Onium Salt Based Systems 318
12.5.2.1	Radical Source/Iodonium Salt Two-Component Systems 318
12.5.2.2	Radical Source/Iodonium Salt Three-Component Systems 320
12.5.2.3	Role of the Onium Salt 329
12.5.2.4	One-Component Radical Source-Onium Salt System 330
12.5.2.5	Photoinitiation Step 330
12.5.2.6	Recent Applications 331
12.5.3	Radical/Metal-Salt-Based Systems 331
12.5.4	Addition/Fragmentation Reaction 331
12.6	Miscellaneous Systems 332
12.7	Photosensitive Systems for Living Cationic Polymerization 332
12.8	Photosensitive Systems for Hybrid Cure 333
	References 333
13	Anionic Photoinitiators 343
13.1	Inorganic Complexes 343
13.2	Organometallic Complexes 343
13.3	Cyano Derivative/Amine System 344
13.4	Photosensitive Systems for Living Anionic Polymerization 344
	References 345
14	Photoacid Generators (PAG) Systems 347
14.1	Iminosulfonates and Oximesulfonates 347
14.2	Naphthalimides 348
14.3	Photoacids and Chemical Amplification 349
	References 349
15	Photobase Generators (PBG) Systems 351
15.1	Oxime Esters 351
15.2	Carbamates 351
15.3	Ammonium Tetraorganyl Borate Salts 351

XVI	Contents	
	15.4	N-Benzylated-Structure-Based Photobases 352
	15.5	Other Miscellaneous Systems 353
	15.6	Photobases and Base Proliferation Processes 354
		References 354
		Part IV Reactivity of the Photoinitiating System 357
	16	Role of the Experimental Conditions in the Performance of a Radical Photoinitiator 359
	16.1	Role of Viscosity 360
	16.2	Role of the Surrounding Atmosphere 361
	16.3	Role of the Light Intensity 361
		References 364
	17	Reactivity and Efficiency of Radical Photoinitiators 367
	17.1	Relative Efficiency of Photoinitiators 367
	17.1.1	Photopolymerization of MMA in Organic Solvents 367
	17.1.2	Photopolymerization of Various Monomers in Organic Solvents and Bulk Media 367
	17.1.3	Photopolymerization of TMPTA in Film 371
	17.1.4	Photopolymerization of Monomer/Oligomer Film 373
	17.1.5	Photopolymerization of Acrylamide in Water 376
	17.1.6	Photopolymerization of Oil-Soluble Monomers in Direct Micelles 376
	17.1.7	Photopolymerization of Water-Soluble Monomers in Reverse Micelles 377
	17.2	Role of the Excited-State Reactivity 377
	17.2.1	Generation of Radicals 377
	17.2.2	Role of Monomer Quenching 378
	17.3	Role of the Medium on the Photoinitiator Reactivity 381
	17.3.1	Reactivity in Solution 381
	17.3.2	Reactivity in Microheterogeneous Solution 383
	17.3.3	Reactivity in Bulk 386
	17.4	Structure/Property Relationships in Photoinitiating Systems 388
	17.4.1	Role of the Bond Dissociation Energy in Cleavable Systems 388
	17.4.2	Role of the Bond Dissociation Energy in Noncleavable Systems 389
	17.4.3	Role of the Initiating Radical in the Initiation Step 391
	17.4.3.1	Generation of the First Monomer Radical 391
	17.4.3.2	Addition Rate Constants to Monomer 392
	17.4.3.3	Description of the Radical/Monomer Reactivity 392
		References 394

18	Reactivity of Radicals toward Oxygen, Hydrogen Donors, Monomers, and Additives: Understanding and Discussion 399
18.1	Alkyl and Related Carbon-Centered Radicals 399
18.2	Aryl Radicals 401
18.3	Benzoyl Radicals 402
18.4	Acrylate and Methacrylate Radicals 403
18.5	Aminoalkyl Radicals 404
18.5.1	Reactivity 404
18.5.2	Role of the Class of the Amine 408
18.5.3	N-Phenyl Glycine Derivatives 408
18.5.4	Chain Length Effect 409
18.5.5	Regioselectivity of the Hydrogen Abstraction Reaction 409
18.5.6	Aminoalkyl Radicals and the Halogen Abstraction
	Reaction 410
18.5.7	Reactivity under Air 411
18.6	Phosphorus-Centered Radicals 412
18.7	Thiyl Radicals 413
18.8	Sulfonyl and Sulfonyloxy Radicals 417
18.9	Silyl Radicals 418
18.9.1	The Particular Behavior of the Tris(Trimethylsilyl)Silyl Radical 421
18.9.2	Reactivity and Photoinitiation under Air 422
18.9.3	Other Sources of Silyl Radicals 424
18.10	Oxyl Radicals 425
18.11	Peroxyl Radicals 426
18.11.1	Interaction with H-Donors 428
18.11.2	Interaction with Monomers 430
18.11.3	Interaction with Triphenylphosphine 430
18.11.4	S _H 2 Substitution 430
18.11.5	Other Oxyls and Peroxyls 431
18.12	Aminyl Radicals 431
18.13	Germyl and Stannyl Radicals 432
18.13.1	Reactivity 433
18.13.2	Reactivity under Air 434
18.13.3	Reactivity and Structure of (TMS) ₃ Ge [•] versus (TMS) ₃ Si [•] 434
18.14	Boryl Radicals 435
18.14.1	Reactivity 436
18.14.2	Reactivity under Air 438
18.14.3	Photoinitiation under Air 438
18.15	Lophyl Radicals 439
18.16	Iminyl Radicals 439
18.17	Metal-Centered Radicals 440
18.18	Propagating Radicals 442
18.19	Radicals in Controlled Photopolymerization Reactions 443
18.19.1	Photoiniferters and Dithiocarbamyl Radicals 443
18.19.2	Light-Sensitive Alkoxyamines and Generation of Nitroxides 445

XVIII	Contents	
	18.20	Radicals in Hydrosilylation Reactions 446 References 447
	19	Reactivity of Radicals: Towards the Oxidation Process 455
	19.1	Reactivity of Radicals toward Metal Salts 455
	19.2	Radical/Onium Salt Reactivity in Free-Radical-Promoted Cationic Photopolymerization 456
		References 459
		Conclusion 461
		Index 465

Abbreviations

AA acrylamide
ABE allylbutylether
ABP aminobenzophenone
ADD acridinediones

AH electron/proton donor
AIBN azo bis-isobutyro nitrile

ALD Aldehydes ALK Alkoxyamines AN acrylonitrile

AOT bis-2 ethyl hexyl sodium sulfosuccinate

APG alkylphenylglyoxylates

AQ anthraquinone

ATR Attenuated reflectance

ATRP atom transfer radical polymerization

BA butylacrylate

BAC [2-oxo-1,2-di(pheny)ethyl]acetate BAPO Bis-acyl phosphine oxide BBD Benzoyl benzodioxolane

BBDOM bisbenzo-[1,3]dioxol-5-yl methanone

BC borane complexes
BD Benzodioxinone

BDE bond dissociation energy

BE benzoin esters

BIP-T bis-(4-tert-butylphenyl) iodonium triflate

BMA butylmethacrylate
BME benzoin methyl ether

BMS benzophenone phenyl sulfide

BP benzophenone BPO benzoyl peroxide

BPSK 1-Propanone,1-[4-[(4-benzoylphenyl)thio]phenyl]-2-methyl-2-[(4-benzoylphenyl)thio]phenyl]

methylphenyl)sulfonyl]

BTTB 4-Benzoyl(4'-tert-butylperoxycarboxyl) tert-butylperbenzoate

BVE butylvinylether

Bz benzil BZ benzoin

C1 7-diethylamino-4-methyl coumarin

C6 3-(2′-benzothiazoryl)-7-diethylaminocoumarin

CA cyanoacrylates
CD cyclodextrin

CIDEP chemically induced electron polarization CIDNP chemically induced nuclear polarization

CL caprolactone

CNT photopolymerized lipidic assemblies

co-I co-initiator

CPG cyano N-phenylglycine CQ camphorquinone CT charge transfer

CTC charge transfer complex
CTP computer-to-plate
CTX chlorothioxanthone
CumOOH cumene hydroperoxide
CW continuous-wave
DB deoxybenzoin

DCPA dicylopentenyl acrylates
DDT diphenyldithienothiophene
DEAP 2,2-dietoxyacetophenone

DEDMSA N,N-diethyl-1,1-dimethylsilylamine DEEA 2-(2-ethoxy-ethoxy) ethyl acrylate

DFT density functional theory

DH hydrogen donor

DMAEB dimethylamino ethyl benzoate

DMPA 2,2-dimethoxy -2 phenyl-acetophenone DMPO 5,5'-dimethyl-1-pyrroline N-oxide

DPA diphenyl acetylene

dPI difunctional photoinitiators
DSC differential scanning calorimetry
DTAC dodecyl trimethylammonium chloride

DUV deep UV
DVE divinylether
EA electronic affinity

EAB diethyl amino benzophenone EDB ethyl dimethylaminobenzoate

EHA 2-ethyl hexyl ester EL ethyl linoleate

EMP N-ethoxy-2-methylpyridinium
EMS epoxy-modified silicone

Eo Eosin Y EP epoxy acrylate EpAc epoxy acrylate; see Section 16 p359 **EPDM** ethylene-propylene-diene monomers **EPHT** electron/proton hydrogen transfer

EPOX 3,4-epoxycyclohexane) methyl 3,4-epoxycyclohexylcarboxylate

EPT ethoxylated pentaerythritol tetraacrylate

exposure reciprocity law ERL. epoxidized soybean oil **ESO ESR** electron spin resonance

ESR-ST Electron spin resonance spin trapping

ETenergy transfer eТ electron transfer EtBz ethylbenzene EUV extreme UV **EVE** ethylvinylether FBs fluorescent bulbs

Fc(+)ferrocenium salt derivative free radical photopolymerization **FRP**

free-radical-promoted cationic polymerization FRPCP

FTIR Fourier transform infrared

FU fumarate gradient index GRIN

HABI 2,2',4,4',5,5'-hexaarylbiimidazole Hindered amine light stabilizer HALS

HAP 2-hydroxy-2- methyl-1- phenyl-1- propanone 1-hydroxy- cyclohexyl-1- phenyl ketone **HCAP**

hydrocarbons **HCs**

HDDA hexane diol diacrylate hfc hyprefine splitting **HFS** hyperfine splitting

HOMO highest occupied molecular orbital hydroquinone methyl ether **HOME**

Highly reactive acrylate monomers **HRAM**

HSG hybrid sol-gel HThydrogen transfer ΙP ionization potential

Interpenetrating polymer networks **IPNs**

infrared IR

ISC intersystem crossing ITX isopropylthioxanthone IAW julolidine derivative

K-ESR kinetic electron spin resonance

KC ketocoumarin

2K-PUR two-component polyurethane LAT light absorbing transients

LCAO linear combination of atomic orbitals LCD liquid crystal display
LDI Laser direct imaging
LDO limonene dioxide
LED light-emitting diode
LFP laser flash photolysis

LIPAC laser-induced photoacoustic calorimetry

LS light stabilizers

LUMO lowest unoccupied molecular orbital

MA methylacrylate MA monomer acceptor

MAL maleate

MB methylene blue

MBI mercaptobenzimidazole
MBO mercaptobenzoxazole
MBT mercaptobenzothiazole
MD monomer donor
MDEA methyldiethanolamine
MDF medium-density fiber
MEK methyl ethyl ketone

MIR multiple internal reflectance

MK Mischler's ketone MMA methylmethacrylate MO molecular orbitals

mPI Multifunctional photoinitiators

MPPK 2-benzyl-2-dimethylamino-1-(4-morpholinophenyl)-1-butanone

MWD molecular weight distribution

NAS 2-[p-(diethyl-amino)styryl]naphtho[1,2-d]thiazole

NHC N-heterocyclic carbene

NIOTf N-(trifluoromethanesulfonyloxy)-1,8-naphthalimide

NIR near-IR reflectance

NMP2 nitroxide-mediated photopolymerization NMP nitroxide-mediated photopolymerization

NMR Nuclear magnetic resonance

NOR norbornenes
NP nanoparticles
NPG N-phenyl glycine
NQ naphthoquinone

NVET nonvertical energy transfer

NVP *N*-vinylpyrolidone OD optical density

OLED organic light-emitting diode
OMC organometallic compounds
On+ onium salt derivative
OrM organic matrixes
P+ pyrilium salt derivative

PAG Photoacid generators PRG photobase generator PBN phenyl-N-tertbutyl nitrone

PC. photocatalyst

PCBs printed circuit boards **PCL** polycaprolactone

1-phenyl 2-propanedione-2 (ethoxycarbonyl) oxime PDO

PEG polyethyleneglycol **PES** potential energy surface PETA pentaerythritol tetraacrylate

PHS Poly(hydrosilane)s PHT pure hydrogen transfer

PΙ photoinitiator

PIS photoinitiating system

PLA Polylactic acid

PI.P pulsed laser polymerization

PI.P Pulsed laser-induced polymerization

PMK 2-methyl-1-(benzoyl)-2-morpholino-propan-1-one

PMMA polymethylmethacrylate POH phenolic compounds PPD 1-phenyl-1,2-propanedione

2-benzyl-2-dimethylamino-1-(phenyl)-1-butanone PPK

PS photosensitizer

PS/PI photosensitizer/photoinitiator Pressure-sensitive adhesives **PSAs**

PVC polyvinylchloride printed wiring boards **PWBs** PYR pyrromethene

RAFT reversible addition-fragmentation transfer

RBRose Bengal

RFID radiofrequency identification

Ring-opening metathesis photopolymerization ROMP

ROOH peroxide derivative ROOH hydroperoxide derivative ROP ring-opening polymerization

radical pair RΡ

RPM radical pair mechanism

RSH mercaptan

real-time Fourier transform infrared RT-FTIR **SCM** solvatochromic comparison method

SCRP spin-correlated radical pair SDS sodium dodecyl sulfate

SG1 N-(2-methylpropyl)-N-(1-diethylphosphono-2,

2-dimethylpropyl)-N-oxyl

SHOMO singly highest occupied molecular orbital SOMO singly occupied molecular orbital

STY styrene SU suberone

SWNT single-wall carbon nanotube

TEA triethyl amine

TEMPO 2,2,6,6, tetramethylpiperidine N-oxyl radical

THF tetrahydrofuran ThP thiophene

TI titanocene derivative

TIPNO 2,2,5-tri-methyl-4-phenyl-3-azahexane-3-nitroxide

TLS thermal lens spectroscopy
TM triplet mechanism

TMP 2,2,6,6-tetramethylpiperidine
TMPTA trimethylolpropane triacrylate
TP+ thiopyrilium salt derivative
TPA two-photon absorption

TPGDA tetrapropyleneglycol diacrylate
TPK 1-[4-(methylthio) phenyl]-ethanone

TPMK 2-methyl-1-(4-methylthiobenzoyl)-2-morpholino-propan-1-one

TPO 2,4,6-trimethyl benzoyl-diphenylphosphine oxide

TPP triphenylphosphine

TR-ESR time resolved electron spin resonance
TR-FTIR time-resolved Fourier transform infra red
TR-S2FTIR Laser-induced step-scan FTIR spectroscopy

TS transition state
TST transition state theory
TTMSS tris(trimethyl)silylsilane

TX thioxanthone

TX-SH 2-mercaptothioxanthone Tz triazine derivative

ULSI ultra large scale integration

UV ultraviolet
UVA UV absorbers
VA vinyl acetate
VC vinylcarbazole
VE vinyl ethers
VE vinylacetate

VET vertical energy transfer

Vi violanthrone VIE vitamin E

VLSI very large scale integration VOC volatile organic compounds

VP vinylpyrrolidone VUV vacuum ultraviolet

XT xanthones

Introduction

Light-induced polymerization reactions are largely encountered in many industrial daily life applications or in promising laboratory developments. The basic idea is to readily transform a liquid monomer (or a soft film) into a solid material (or a solid film) on light exposure. The huge sectors of applications, both in traditional and high-tech areas, are found in UV curing (this area corresponds to the largest part of radiation curing that includes UV and electron beam curing), laser imaging, microlithography, stereolithography, microelectronics, optics, holography, medicine, and nanotechnology.

UV curing represents a green technology (environmentally friendly, nearly no release of volatile organic compounds (VOCs), room temperature operation, possible use of renewable materials, use of convenient light sources (light-emitting diodes (LEDs), household lamps, LED bulbs, and the sun) that continues its rapid development. The applications concern, for example, the use of varnishes and paints (for a lot of applications on a large variety of substrates, e.g., wood, plastics, metal, and papers), the design of coatings having specific properties (for flooring, packaging, release papers, wood and medium-density fiber (MDF) panels, automotive, pipe lining, and optical fibers), the development of adhesives (laminating, pressure sensitive, and hot melt), and the graphic arts area (drying of inks, inkjets, overprint varnishes, protective and decorative coatings, and the manufacture of conventional printing plates).

Other applications of photopolymerization reactions concern medicine (restorative and preventative denture relining, wound dressing, ophthalmic lenses, glasses, artificial eye lens, and drug microencapsulation), microelectronics (soldering resists, mask repairs, encapsulants, conductive screen inks, metal conductor layers, and photoresists), microlithography (writing of complex relief structures for the manufacture of microcircuits or the patterning of selective areas in microelectronic packaging using the laser direct imaging (LDI) technology; direct writing on a printing plate in the computer-to-plate technology), 3D machining (or three-dimensional photopolymerization or stereolithography) that gives the possibility of making objects for prototyping applications, optics (holographic recording and information storage, computer-generated and embossed holograms, manufacture of optical elements, e.g., diffraction grating, mirrors, lenses, waveguides, array illuminators, and display devices), and structured materials on the nanoscale size.

Photopolymerization reactions are currently encountered in various experimental conditions, for example, in film, gas phase, aerosols, multilayers, (micro)heterogeneous media or solid state, on surface, in ionic liquids, in situ for the manufacture of microfluidic devices, in vivo, and under magnetic field. Very different aspects can be concerned with gradient, template, frontal, controlled, sol-gel, two-photon, laser-induced or spatially controlled, and pulsed laser photopolymerization.

As a photopolymerization reaction involves a photoinitiating system, a polymerizable medium, and a light source, a strong interplay should exist between them. The photoinitiator has a crucial role as it absorbs the light, converts the energy into reactive species (excited states, free radicals, cations, acids, and bases) and starts the reaction. Its reactivity governs the efficiency of the polymerization. A look at the literature shows that a considerable number of works are devoted to the design of photosensitive systems being able to operate in many various (and sometimes exotic) experimental conditions. This research field is particularly rich. Fantastic developments have appeared all along the past three decades. Significant achievements have been made since the early works on photopolymerization in the 1960s and the traditional developments of the UV-curing area. At present, high-tech applications are continuously emerging. Tailor-made photochemistry and chemistry have appeared in this area. The search for a safe and green technology has been launched. Interesting items relate not only to the polymer science and technology field but also to the photochemistry, physical chemistry, and organic chemistry areas.

We believe that the proposed book focused on this exciting topic related to the photosensitive systems encountered in photopolymerization reactions will be helpful for many readers. Why a new book? Indeed, in the past 20 years, many aspects of light-induced polymerization reactions have been obviously already discussed in books and review papers. Each of these books, however, usually covers more deeply selected aspects depending first on the origin (university, industry) and the activity sector of the author (photochemistry, polymer chemistry, and applications) and second on the goals of the book (general presentation of the technology, guide for end users, and academic scope). Our previous general book published more than 15 years ago (1995) and devoted to the three photoinitiation-photopolymerization-photocuring complementary aspects already provided a first account on the photosensitive systems.

For obvious reasons, all these three fascinating aspects that continuously appear in the literature cannot be unfortunately developed now (in 2011) in detail in a single monograph because of the rapid growth of the research. A book that mostly concentrates on the photosensitive systems that are used to initiate the photopolymerization reaction, their adaptation to the light sources, their excited state processes, the reactivity of the generated initiating species (free radicals, acids, and bases), their interaction with the different available monomers, their working out mechanisms, and the approach for a complete understanding of the (photo)chemical reactivity was missing. This prompted us to write the present book. It aims at providing an original and up-to-date presentation of these points together with a discussion of the structure/reactivity/efficiency relationships observed in photoinitiating systems usable in radical, cationic, and anionic photopolymerization as well as in acid and base catalyzed photocrosslinking reactions. We wish to focus on the necessary role of the basic research toward the progress of the applied research through the large part we have devoted to the involved mechanisms. In fact, everybody is aware that there is no real technical future development without a present high-quality scientific research. In our opinion, such an extensive and complete book within this philosophy has never been written before.

Science is changing very fast. During the preparation of a book, any author has the feeling of walking behind the developments that unceasingly appear. It is rather difficult to have the latest photography of the situation by the end of the manuscript; this is also reinforced by the necessary delay to print and deliver the book. Therefore, we decided here to give not only the best up-to-date situation of the subject but also to take time to define a lot of basic principles and concepts, mechanistic reaction schemes, and examples of reactivity/efficiency studies that remain true and are not submitted to a significant aging on a 10-year timescale.

The book is divided into four parts. In Part I, we deliver a general presentation of the basic principles and applications of the involved photopolymerization reactions with a description of the available light sources, the different monomers and the properties of the cured materials, the various aspects and characteristics of the reactions, and the role of the photosensitive systems and the typical examples of applications in different areas. The part especially concerned with the polymer science point of view (as other books have already dealt in detail with this aspect) focuses on general considerations and latest developments and to what is necessary to clearly understand the following parts. Then, we enter into the heart of the book.

In Parts II and III, we give (i) the most exhaustive presentation of the commercially and academically used or potentially interesting photoinitiating systems developed in the literature (photoinitiators, co-initiators, photosensitizers, macrophotoinitiators, multicomponent combinations, and tailor-made compounds for specific properties), (ii) the characteristics of the excited states, and (iii) the involved reaction mechanisms. We provide an overview of all the available systems but we focus our attention on newly developed photoinitiators, recently reported studies, and novel data on previous well-known systems. All this information is provided for radical photopolymerization (Part II) and cationic and anionic photopolymerization and photoacid and photobase catalyzed photocrosslinking (Part III).

In Part IV, we gather and discuss (i) a large set of data, mostly derived from time-resolved laser spectroscopy and electron spin resonance (ESR) experiments, related to both the photoinitiating system excited states and the initiating radicals (e.g., a complete presentation of the experimental and theoretical reactivity of more than 15 kinds of radicals is provided); (ii) the most recent results of quantum mechanical calculations that allow probing of the photophysical/photochemical properties as well as the chemical reactivity of a given photoinitiating system; and (iii) the reactivity in solution, in micelle, in bulk, in film, under air, in low viscosity media, or under low light intensities.

The book also outlines the latest developments and trends for the design of novel molecules. This concerns first the elaboration of smart systems exhibiting well-designed functional properties or/and suitable for processes in the nanotechnology area. A second direction refers to the development of an evergreen (photo)chemistry elaborating, for example, safe, renewable, reworkable, or biocompatible materials. A third trend is related to the use of soft irradiation conditions for particular applications, which requires the design of low oxygen sensitivity compounds under exposure to low-intensity visible light sources, sun, LEDs, laser diodes, or household lamps (e.g., fluorescence or LED bulbs).

When questioning the Chemical Abstract database, many references appear. We have not intended to give here an exhaustive list of references or a survey of the patent literature. We used, however, more than 2000 references. Pioneer works are cited but our present list of references mainly refers to papers dispatched during the past 15 years. The selection of the articles is most of the time a rather hard and sensitive task. We have done our best and beg forgiveness for possible omissions.

This research field has known a fantastic evolution. We would like now to share the real pleasure we had (and still have) in participating and contributing to this area. Writing this book was really a great pleasure. We hope that our readers, R and D researchers, engineers, technicians, University people, and students involved in various scientific or/and technical areas such as photochemistry, polymer chemistry, organic chemistry, radical chemistry, physical chemistry, radiation curing, imaging, physics, optics, medicine, nanotechnology will appreciate this book and enjoy its content.

And now, it is time to dive into the magic of the photoinitiator/photosensitizer world!